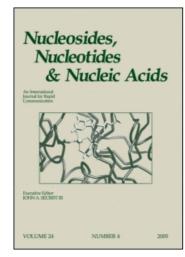
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New Neplanocin Analogues. V. A Potent Adenosylhomocysteine Hydrolase Inhibitor Lacking Antiviral Activity. Synthesis And Antiviral Activity Of 6"-Carboxylic Acid Derivatives Of Neplanocin A

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NEW NEPLANOCIN ANALOGUES. V. A POTENT ADENOSYLHOMOCYSTEINE HYDROLASE INHIBITOR LACKING ANTIVIRAL ACTIVITY. SYNTHESIS AND ANTIVIRAL ACTIVITY OF 6'-CARBOXYLIC ACID DERIVATIVES OF NEPLANOCIN A¹

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Abstract: The 6'-carboxylic acid derivative of neplanocin A 3 was synthesized from NPA, and was converted to the corresponding methyl ester 4 and amides 5 and 6. These were evaluated for their anti-RNA-virus activities. Of the derivatives synthesized, only 5 was active against RNA viruses within the concentration range of 0.14-4.88 µg/mL. Compounds 3 and 5 showed a potent inhibitory effect on S-adenosylhomocysteine (AdoHcy) hydrolase from rabbit erythrocytes. Although a close correlation between the inhibitory effect of adenosine analogues on AdoHcy hydrolase and their antiviral potency has been demonstrated, 3 did not show any anti-RNA-virus activities.

Introduction

The enzyme S-adenosylhomocysteine (AdoHcy) hydrolase, which is important in the regulation of S-adenosylmethionine-dependent transmethylation, has been recognized as an important target for broad-spectrum antiviral agents. Use transmethylation reactions are involved in the maturation of viral mRNAs and hence play a critical role in the virus replication cycle. In fact, a close correlation has been demonstrated between the inhibitory effects of adenosine (Ado) analogues on AdoHcy hydrolase and their antiviral potency.

Neplanocin A (1, NPA),⁶ a carbocyclic nucleoside antibiotic, is the most potent AdoHcy hydrolase inhibitor,⁷ and also has a strong antiviral activity against various RNA and DNA viruses.⁸ However, NPA itself is cytotoxic,⁹ and is inactivated by Ado

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deaminase soon after administration.6,8,10a It has been recognized that the detrimental toxicity of NPA could be derived, for the most part, from phosphorylation of the primary hydroxyl group at the 6'-position (corresponding to the 5'-hydroxyl group of Ado) by Ado kinase and subsequent metabolism by cellular enzymes.⁹ Consequently, to remove or reduce such side effects, much attention has been focused on chemical modification of NPA by us¹⁰ and others.¹¹

It was reported that the 3'-hydroxyl group of NPA is oxidized by NAD+ bound to AdoHcy hydrolase to form an oxidized intermediate 2, that has an enone structure, and this intermediate 2 binds tightly to the enzyme (FIG. 1).¹² However, the 3'-enone derivative 2 could not be isolated due to its instability.¹²

There is a possibility that the derivatives having an enone structure might bind tightly to the enzyme. Therefore, we designed NPA analogues having an enone system, namely 6'-carboxylic acid (3, 9-[(1R,2S,3R)-2,3-dihydroxy-4-carboxy-4-cyclopenten-1-yl]adenine) and its derivatives 4, 5, and 6 as potent AdoHcy hydrolase inhibitors, which could be considered as a kind of constitutional isomer of 2 (FIG. 2). Moreover, it may be possible that designed derivatives 3, 4, 5, and 6 could serve as Michael acceptors of the nucleophilic functional groups of the enzyme side chains.

Chemistry

The synthetic route for 6'-carboxylic acid 3 and its derivatives is outlined in FIG. 3. An attempt to oxidize the 6'-hydroxymethyl moiety of N⁶-benzoyl-2,3-O-isopropylidene-NPA (7) to a carboxyl group directly by KMnO₄ was unsuccessful. Therefore we used a stepwise oxidation of the 6'-hydroxymethyl moiety for the preparation of 3. First, 7 was oxidized to 6'-aldehyde 8 with BaMnO₄, 10a and it was further oxidized by a H₂NSO₃H/NaClO₂ system to afford 6'-carboxylic acid 9 in 68% yield from 7. Deprotection of 9 was done by successive treatment with 25% NH₄OH and 1 N HCl to afford the desired carboxylic acid 3 in 86% yield. Esterification of 3 with HCl/MeOH was done to give methyl ester 4 in 42% yield.

Next, preparation of amide derivatives 5 and 6 were examined. Ammonolysis of 4 with NH₃/MeOH was unsuccessful, because of concomitant formation of multiple byproducts. On the other hand, in the case of a reaction of 4 with MeNH₂ in EtOH, an undesired Michael addition reaction on the enone system of 4 proceeded in preference to ammonolysis of the ester moiety. When the reaction was done at 0 °C for 1 h, 10a and 10b were obtained in 75% and 15% yields, respectively, though 10b was obtained as a sole product in 82% yield by a prolonged reaction period. This indicated that the Michael addition reaction occurred first, followed by the amide formation. In this case, the Michael addition reaction proceeded from the α face of the cyclopentene-ring in a *cis*

FIG. 1

FIG. 2

manner to give **10a** initially, then it was gradually converted to its *N*-methyl-carbamoyl derivative **10b**.

Structures of the Michael adducts 10a and 10b were identified from the 1H NMR spectra. Nuclear overhauser effect (NOE) difference experiments of 10b done in DMSO- d_6 showed a characteristic increase of 2'-OH, 3'-OH, 1'- and 4'-proton signals when 5'-N-methyl protons were irradiated. Similarly, irradiation on the 2'-proton caused a characteristic increase of the 5'-proton signal as well as 8- and 3'-proton signals as illustrated in FIG. 4. These data indicated that both the 5'-MeNH group and 4'-proton were in the α face (signifying cis addition) of the cyclopentane ring. From a comparison of $J_{4',5'}$ values of 10a and 10b (6.9 Hz and 7.3 Hz, respectively), we concluded that 10a and 10b have the same 4',5'-stereochemistry.

This shows that methylamine selectively attacked at the 5'-position of 4 from the α face because of the steric hindrance by the adenine base, then protonation at the 4'-position occurred from the α face to minimize the steric repulsion of the product.

- a) BaMnO₄, CHCl₃; b) H₂NSO₃H, NaClO₂, Acetone; c) 25% NH₄OH; d) 1 N HCl;
- e) HCI, MeOH; f) CICOOBu^I, Et₃N, DMF; g) 30% MeNH₂ In EtOH; h) NH₃ gas;
- i) reflux in MeOH

Fig. 3

NOE interactions of 10b.

FIG. 4

These results suggested that activation of the carboxyl moiety would be necessary to make the ammonolysis proceed prior to the Michael addition reaction. Therefore, the mixed acid-anhydride method was used. To prepare the amide derivatives 5 and 6, 6'-carboxylic acid 9 was converted to its mixed acid-anhydride by treating it with Et₃N and ClCOOBuⁱ in DMF, and then it was treated with anhydrous NH₃ or MeNH₂ to give the corresponding amide 11 (54% yield) or N-methylamide 12 (60% yield), respectively. Removal of N-benzoyl groups of 11 and 12 with the usual NH₃/MeOH conditions were unsuccessful, probably due to the unstable enone system. However, when 11 and 12 were heated under reflux in MeOH, the N-benzoyl groups were deprotected to give desired 5 (85% yield) and 6 (80% yield), respectively.

Biological activity

The susceptibilities of the derivatives to Ado deaminase from calf intestine were studied together with NPA, and the results are shown in FIG. 5. Under the conditions used, NPA itself was deaminated rapidly into a biologically inactive inosine congener, but all of the other compounds tested were completely resistant to calf intestinal Ado deaminase.

Next, we evaluated the inhibitory effects on AdoHcy hydrolase from rabbit erythrocytes and antiviral activities *in vitro*, together with cytotoxicity against Vero cells. The biological results are summarized in Table 1. None of the compounds tested except for NPA were cytotoxic to host cells (Vero) in stationary phase at a concentration up to 500 µg/mL. Of the derivatives, only amide derivative 5 had anti-RNA-virus activities against parainfluenza-3 (PINF-3), measles, or mumps viruses within the EC₅₀ range of

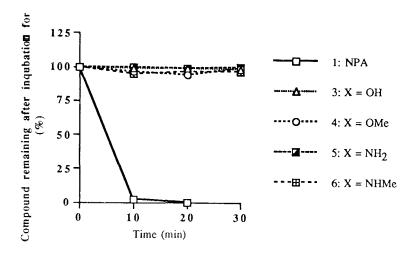


FIG. 5

Effects of calf intestinal adenosine deaminase on NPA and its derivatives.

Table 1. Antiviral activities, a cytotoxicity on Vero cells, and inhibitory effects on AdoHcy hydrolase of NPA and its derivatives

		EC ₅₀ ^b (μg/mL)			IC ₅₀ (μg/mL)	
Compound		PIFV-3	Measles	Mumps	Vero cellc	AdoHcy
	(X)	(C 243)	(Sugiyama)	(ECXH3)	(stationary)	hydrolased
NPA		0.39	0.10	0.11	152e	0.004f
3	(OH)	>25	>25	>25	>500	0.33
4	(OMe)	>25	>25	>25	>500	>50
5	(NH_2)	4.88	2.72	0.14	>500	2.56
6	(NHMe)	>25	>25	>25	>500	>50
10a		>100	NDg	ND	ND	ND
10b		>100	ND	ND	ND	ND

^a Antiviral assay was done by a previously reported method. ^{10b}

b Concentration required to inhibit virus-induced cytopathogenicity by 50%.

^c MTT method was used to measure cytotoxicity.

d From rabbit erythrocytes; Assay was done by a previously reported method. 10a

e Data was taken by Ref. 10d. f Data was taken by Ref. 10a.

g Not determined.

0.14- $4.88 \mu g/mL$. 6'-Carboxylic acid derivative 3 and amide derivative 5 were potent inhibitors of AdoHcy hydrolase from rabbit erythrocytes (IC₅₀ = $0.33 \mu g/mL$, 2.56 $\mu g/mL$, respectively). Although a close correlation between the inhibitory effect of adenosine analogues on AdoHcy hydrolase and their antiviral potency has been demonstrated, 4 6'-carboxylic acid derivative 3 did not show any anti-RNA-virus activities in spite of its significant inhibitory effect on AdoHcy hydrolase.

To the best of our knowledge, 3 would be the first adenosine analogue that is inactive against viruses, but inhibits AdoHcy hydrolase significantly. This may be due to its inability to pass through the cell membrane, because of its high polarity. Compound 4 did not show any inhibitory effect on AdoHcy hydrolase, in spite of having an enone structure and an ability to act as a Michael acceptor.

Experimental section

Melting points were measured on a Yanagimoto MP-3 micromelting point apparatus and are uncorrected. NMR spectra were recorded with a JEOL FX-270, or a GCX-400 spectrometer with tetramethylsilane as an internal standard. Mass spectra were measured on a JEOL SX-102 spectrometer. High-resolution mass spectra were measured on a JMX DX-303 spectrometer. Thin-layer chromatography was done on E. Merck precorted plates 60F₂₅₄. Flash chromatography was conducted with E. Merck silica gel 9385. NPA were prepared according to a reported method.⁶

N⁶-Benzoyl-9-[(1R,2S,3R)-2,3-O-isopropylidenedioxy-4-carboxy-4-cyclopenten-1-yl]adenine (9). A mixture of 7 (1 g, 2.46 mmol) and BaMnO₄ (10 g, 39.00 mmol) in CHCl₃ (100 mL) was stirred at 65 °C for 5 h. The oxidizing agent was filtered off and the filter was washed with hot CHCl₃. The filtrate was evaporated to dryness under reduced pressure. The residue was dissolved in acetone (100 mL), and then 0.3 N H₂NSO₃H (17.6 mL) was added dropwise over 5 min at -30 °C. Three minutes later, 0.2 N NaClO₂ (14.7 mL) was added dropwise over 3 min, and the mixture was stirred at -10 °C for 30 min. To the reaction mixture, cooled water (4 °C, 100 mL) was added, and the resulting mixture was extracted with CHCl₃ (100 mL×3). The combined CHCl₃ phase was filtered through Whatman 1PS filter paper and evaporated. The residue was purified by flash chromatography (silica gel, CHCl₃/MeOH, 5:1 followed by CHCl₃/MeOH/H₂O, 65:25:3) to give 707 mg (68%) of 9 as a solid: ¹H NMR (270 MHz, CDCl₃/CD₃OD, 8:1) δ 8.78 and 8.10 (each s, each 1 H, H-8 and 2), 8.09 (d, 2 H, H-Ph, J = 6.9 Hz), 7.62 (dd, 1 H, H-Ph, J = 7.3, 7.3 Hz), 7.56 (dd, 2 H, H-Ph, J = 7.3, 6.9 Hz), 6.61 (br s, 1 H, H-5'), 5.79 (br s, 1 H, H-1'), 5.72 (m, 1 H, H-3'), 4.87 (br d, 1 H, H-2', J = 5.3 Hz), 1.54, 1.41 (each s, each 3 H, isopropyl-CH₃); MS (FAB positive) m/z 422 (MH+); HRMS (FAB positive), calcd for C₂₁H₂₀O₅N₅ (MH⁺) 422.1464, found 422.1456.

9-[(1R,2S,3R)-2,3-Dihydroxy-4-carboxy-4-cyclopenten-1-yl]adenine

hydrochloride (3). A solution of 9 (300 mg, 0.71 mmol) in 25% NH₄OH (10 mL) was stirred at 45 °C for 1 h. The solvent was removed under reduced pressure. MeOH (5 mL) was added to the residue, and the mixture was evaporated. The residue was dissolved in 1 N HCl (5 mL), and the solution was stirred at room temperature for 1 h. The solvent was removed under reduced pressure, and then H₂O (5 mL) was added to the residue, and the mixture was evaporated. The residue was purified on a Dowex-1 ion-exchange resin column (1×5 cm, OH⁻ form, H₂O \rightarrow 50% MeOH \rightarrow 0.1 N HCl). The fractions containing the desired compound were evaporated to dryness. The residue was dissolved in a small amount of MeOH, then Et₂O was added to give 191 mg (86%) of 3 as a white precipitate: ¹H NMR (270 MHz, CD₃OD) δ 8.40 and 8.38 (each s, each 1 H, H-8 and 2), 6.97 (d, 1 H, H-5', J = 1.7 Hz), 5.75 (br d, 1 H, H-1', J = 5.6, 1.7 Hz), 4.84 (d, 1 H, H-3', J = 4.6 Hz), 4.52 (dd, 1 H, H-2', J = 5.6, 6.9 Hz); MS (FAB positive) m/z 278 (MH⁺); HRMS (FAB negative), calcd for C₁₁H₁₀O₄N₅ (M-H)⁻ 276.0733, found 276.0715.

9-[(1R,2S,3R)-2,3-Dihydroxy-4-methoxycarbonyl-4-cyclopenten-1-yl]adenine (4). Acetyl chloride (300 µL) was added to MeOH (3 mL), and the solution was stirred for 30 min. To the solution, 3 (80 mg, 0.25 mmol) was added, and the mixture was stirred at room temperature overnight. The solvent was removed under reduced pressure and coevaporated with MeOH. To the residue, CHCl₃ (10 mL) was added, and the insoluble materials were collected by filtration, which was purified by flash chromatography (silica gel, CHCl₃/MeOH/H₂O, 65:25:3) to give 31 mg (42%) of 4 as a solid. An analytical sample was obtained by crystallization from MeOH: mp > 207 °C (dec.); ¹H NMR (270 MHz, CD₃OD) δ 8.07 and 8.06 (each s, each 1 H, H-8 and 2), 6.90 (d, 1 H, H-5', J = 1.7 Hz), 5.54 (br d, 1 H, H-1', J = 6.9 Hz), 4.82-4.70 (m, H-3' and MeOH), 4.46 (dd, 1 H, H-2', J = 6.9, 5.6 Hz), 3.72 (s, 3 H, CH₃O); MS (FAB positive) m/z 292 (MH⁺). Anal.Calcd for C₁₂H₁₃N₅O₄: C, 49.48; H, 4.50; N, 24.04. Found: C, 49.23; H, 4.30; N, 23.82.

9-[(1R,2S,3R,4S,5S)-2,3-Dihydroxy-4-methoxycarbonyl-5-(N-methylamino)-4-cyclopenten-1-yl]adenine (10a) and 9-[(1R,2S,3R,4S,5S)-2,3-Dihydroxy-4-(N-methylamino)-4-cyclopenten-1-yl]adenine (10b). Method A: Compound 4 (35 mg, 0.12 mmol) was added to 30% MeNH₂ in EtOH (1 mL) at 0 °C, and the resulting mixture was stirred at the same temperature for 1 h. The solvent was removed under reduced pressure, and coevaporated with EtOH. The residue was purified by preparative TLC (silica gel; developed with CHCl₃/MeOH/H₂O, 65:25:3; extracted with CHCl₃/MeOH, 2:1) to give 29 mg (75%) of 10a and 6 mg (15%) of 10b as solids, respectively. Method B: Compound 4 (11 mg, 0.04 mmol) was added to 30% MeNH₂ in EtOH (1.1 mL), and the mixture was stirred at room temperature for 27 h. The reaction mixture was worked up according to the procedure in method A to give 10 mg (82%) of

10b as a solid. Physical data for 10a: 1 H NMR (270 MHz, CD₃OD) δ 8.20 and 8.19 (each s, each 1 H, H-8 and 2), 4.71 (dd, 1 H, H-1', J = 8.9, 8.9 Hz), 4.55 (dd, 1 H, H-2', J = 8.9, 5.3 Hz), 4.38 (dd, 1 H, H-3', J = 5.3, 3.6 Hz), 3.95 (dd, 1 H, H-5', J = 8.9, 6.9 Hz), 3.79 (s, 3 H, CH₃O), 2.89 (dd, 1 H, H-4', J = 6.9, 3.6 Hz), 2.22 (s, 3 H, 5'-CH₃N); MS (FAB positive) m/z 323 (MH+); HRMS (FAB positive), calcd for C₁₃H₁₉N₆O₄ (MH+) 323.1468, found 323.1490. Physical data for 10b: 1 H NMR (270 MHz, CD₃OD) δ 8.32 and 8.20 (each s, each 1 H, H-8 and 2), 4.76 (dd, 1 H, H-1', J = 8.8, 8.3 Hz), 4.49 (dd, 1 H, H-2', J = 8.3, 5.4 Hz), 4.22 (dd, 1 H, H-3', J = 5.4, 4.4 Hz), 3.83 (dd, 1 H, H-5', J = 8.8, 7.3 Hz), 2.80 (s, 3 H, 6'-CH₃N), 2.73 (dd, 1 H, H-4', J = 7.3, 4.4 Hz), 2.23 (s, 3 H, 5'-CH₃N); MS (FAB positive) m/z 322 (MH+); HRMS (FAB positive), calcd for C₁₃H₂₀N₇O₃ (MH+) 322.1627, found 322.1639.

 N^6 -Benzoyl-9-[(1R,2S,3R)-2,3-O-isopropylidenedioxy-4-carbamoyl-4cyclopenten-1-yl]adenine (11). To a mixture of 9 (150 mg, 0.36 mmol) and Et₃N (150 μ L, 1.07 mmol) in DMF (2 mL) at -7 °C, ClCOOBuⁱ (91 μ L, 0.72 mmol) was added and the mixture was stirred for 30 min. Anhydrous NH3 gas was bubbled into the mixture for 7 min, and the resulting mixture was stirred at -5 °C for 2 h, then the mixture was evaporated. To the residue, CHCl₃ (30 mL) and distilled water (5 mL) were added, and the mixture was partitioned. The organic layer was filtered through Whatman 1PS filter paper, and the filtrate was evaporated. The residue was purified by flash chromatography (silica gel, CHCl₃/MeOH, 40:1→30:1) to give 81 mg (54%) of 11 as solids. An analytical sample was obtained by crystallization from ethyl acetate/hexane: mp >146 °C (dec.); ¹H NMR (270 MHz, CDCl₃) δ 8.11 (br s, 1 H, NH), 8.80 and 7.95 (each s, each 1 H, H-8 and 2), 8.02 (d, 2 H, Ph, J = 7.3 Hz), 7.64-7.48 (m, 3 H, Ph), 6.72 (s, 1 H, H-5'), 6.61, 5.87 (each s, each 1 H, NH₂CO), 5.71-5.69 (m, 2 H, H-1', 3'), 4.91 (d, 1 H, H-2', J = 5.9Hz), 1.53, 1.42 (each s, each 3 H, isopropyl-CH₃); MS (FAB positive) m/z 421 (MH+). Anal.Calcd for C₂₁H₂₀N₆O₄•1/2 H₂O: C, 58.74; H, 4.93; N, 19.57. Found: C, 58.95; H, 5.03; N, 19.56.

9-[(1R,2S,3R)-2,3-Dihydroxy-4-carbamoyl-4-cyclopenten-1-yl]adenine (5). A solution of 11 (35 mg, 0.08 mmol) in MeOH (2 mL) was heated under reflux for 52 h. After the mixture cooled to room temperature, 1 N HCl (2 mL) was added to the solution, and the mixture was stirred at room temperature for 5 h. The solvent was removed under reduced pressure, a small amount of MeOH was added to the residue, and then the solvent was evaporated. MeOH (5 mL) and Et₃N (100 μL) was added to the residue, and the mixture was stirred for 5 min, then the resulting mixture was evaporated. The residual solid was collected by filtration and washed with CHCl₃, which was purified by flash chromatography (silica gel, CHCl₃/MeOH, 5:1→3:1 followed by CHCl₃/MeOH /H₂O, 65:25:3) to give 20 mg (85%) of 5 as solids. An analytical sample was obtained by

crystallization from MeOH: mp 194-196 °C; 1 H NMR (270 MHz, CD₃OD) δ 8.19 and 8.14 (each s, each 1 H, H-8 and 2), 6.77 (d, 1 H, H-5', J = 2.0 Hz), 5.61 (br d, 1 H, H-1', J = 6.6 Hz), 4.90 (d, 1 H, H-3', J = 5.6 Hz), 4.53 (dd, 1 H, H-2', J = 5.6, 6.6 Hz); MS (FAB positive) m/z 277 (MH+). Anal.Calcd for C₁₁H₁₂N₆O₃·H₂O: C, 44.90; H, 4.80; N, 28.56. Found: C, 45.20; H, 4.74; N, 28.36.

 N^6 -Benzoyl-9-[(1R,2S,3R)-2,3-O-isopropylidenedioxy-4-(N-methylcarbamoyl)-4-cyclopenten-1-yl]adenine (12). To a mixture of 9 (150 mg, 0.36 mmol) and Et3N (150 μ L, 1.07 mmol) in DMF (1.5 mL) at 0 °C, ClCOOBuⁱ (93 μ L, 0.72 mmol) was added and the mixture was stirred for 10 min. 30% MeNH₂ in EtOH (150 µL) was added to the mixture at -78 °C, and the mixture was stirred for 20 min at the same temperature. To the mixture, saturated NH₄Cl (15 mL) and ethyl acetate (30 mL) were added, and the mixture was partitioned. The organic layer was washed with water, filtered through Whatman 1PS filter paper, and evaporated. The residue was purified by preparative TLC (silica gel; developed and extracted with CHCl₃/MeOH, 10:1) to give 93 mg (60%) of 12 as solids. An analytical sample was obtained by crystallization from ethyl acetate / hexane: mp 149-150 °C; ¹H NMR (270 MHz, CDCl₃) δ 9.11 (br s, 1 H, NH), 8.79 and 7.93 (each s, each 1 H, H-8 and 2), 8.02 (d, 2 H, Ph, J = 7.3 Hz), 7.64-7.48 (m, 3 H, Ph), 6.68 (d, 1 H, H-5', J = 2.6 Hz), 6.65 (br d, 1 H, NHCO, J = 5.0 Hz), 5.69 (br d, 1 H, H-1', J = 2.0 Hz), 5.66 (d, 1 H, H-3', J = 5.6 Hz), 4.89 (d, 1 H, H-2', J = 5.6 Hz), 2.97 (d, 3 H, CH₃N, J =5.0 Hz), 1.52, 1.42 (each s, each 3 H, isopropyl-CH₃); MS (FAB positive) m/z 435 (MH+). Anal.Calcd for C₂₂H₂₂N₆O₄•2/5 H₂O: C, 59.83; H, 5.20; N, 19.03. Found: C, 59.97; H, 4.93; N, 18.84.

9-[(1R,2S,3R)-2,3-Dihydroxy-4-(N-methylcarbamoyl)-4-cyclopenten-1-

yl]adenine (6). A solution of 12 (53 mg, 0.12 mmol) in MeOH (10 mL) was heated under reflux for 74 h. After this cooled to room temperature, 1 N HCl (5 mL) was added to the solution, and the mixture was stirred at room temperature for 1 h. The solvent was removed under reduced pressure, and then distilled water (5 mL) was added to the residue, and the solvent was evaporated. MeOH (5 mL) and Et₃N (100 μ L) was added to the residue, and the mixture was stirred for 5 min, and then the resulting mixture was evaporated. The residual solid was collected by filtration and washed with CHCl₃. The solid obtained was purified by flash chromatography (silica gel, CHCl₃/MeOH, 5:1 \rightarrow 3:1 followed by CHCl₃/MeOH/H₂O, 65:25:3) to give 28 mg (80%) of 6 as solids. An analytical sample was obtained by crystallization from MeOH: mp >224 °C (dec.); ¹H NMR (270 MHz, CD₃OD) δ 8.18 and 8.13 (each s, each 1 H, H-8 and 2), 6.71 (d, 1 H, H-5', J = 2.0 Hz), 5.60 (br d, 1 H, H-1', J = 6.3 Hz), 4.91 (d, 1 H, H-3', J = 5.6 Hz), 4.52 (dd, 1 H, H-2', J = 5.6, 6.6 Hz), 2.86 (s, 3 H, CH₃N); MS (FAB) m/z 278 (MH+). Anal.Calcd for C₁₂H₁₄N₆O₃-1/2 H₂O: C, 48.16; H, 5.05; N, 28.08. Found: C, 48.43; H, 5.09; N, 28.20.

REFERENCES

- 1. This paper constitutes part 146 of Nucleosides and Nucleotides. Part 145. Shuto, S.; Haramuishi, K.; Matsuda, A. *Tetrahedron Lett*. Submitted.
- Ueland, P.M. Pharmacol. Rev. 1982, 34, 223-253.
- 3. De Clercq, E. Biochem. Pharmacol. 1987, 36, 2567-2575.
- 4. Wolfe, M. S.; Borchardt, R. T. J. Med. Chem. 1991, 34, 1521-1530.
- 5. Cools, M.; De Clercq, E. *Biochem. Pharmacol.* **1989**, *38*, 1061-1067.
- a) Yaginuma, S.; Muto, N.; Tsujino, M.; Sudate, Y.; Hayashi, M.; Otani, M. J. Antibiot. 1981, 34, 359-366.
 b) Hayashi, M.; Yaginuma, S.; Yoshioka, H.; Nakatsu, K. J. Antibiot. 1981, 34, 675-680.
- Borchardt, R. T.; Keller, B. T.; Patel-Thombre, U. Neplanocin A. J. Biol. Chem. 1984, 259, 4353-4358.
- 8. De Clercq, E. Antimicrob. Agents Chemother. 1985, 28, 84-89.
- a) Glazer, R. I.; Knode, M. C. J. Biol. Chem. 1984, 259, 12964-12969. b) Inaba, M.; Nagashima, K.; Tsukagoshi, S.; Sakurai, Y. Cancer Res. 1986, 46, 1063-1067. c) Hoshi, A.; Yoshida, M.; Iigo, M.; Tokuzen, R.; Fukukawa, K.; Ueda, T. J. Phamacobio-Dyn. 1986, 9, 202-206.
- a) Shuto, S.; Obara, T.; Toriya, M.; Hosoya, M.; Snoeck, R.; Andrei, G.; Balzarini, J.; De Clercq, E. J. Med. Chem. 1992, 35, 324-331. b) Shigeta, S.; Mori, S.; Baba, M.; Ito, M.; Honzumi, K.; Nakamura, K.; Oshitani, H.; Numazaki, Y.; Matsuda, A.; Obara, T.; Shuto, S.; De Clercq, E. Antimicrob. Agents Chemother. 1992, 36, 435-439. c) Shuto, S.; Obara, T.; Itoh, H.; Kosugi, Y.; Saito, Y.; Toriya, M.; Yaginuma, S.; Shigeta, S.; Matsuda, A. Chem. Pharm. Bull. 1994, 42, 1688-1690. d) Shuto, S.; Obara, T.; Kosugi, Y.; Saito, Y.; Toriya, M.; Yaginuma, S.; Shigeta, S.; Matsuda, A. BioMed. Chem. Lett. 1994, 4, 605-608.
- For examples; a) Borcherding, D. R.; Narayanan, S; Hasobe, M.; McKee, J. G.; Keller, B. T.; Borchardt, R. T. J. Med. Chem. 1988, 31, 1729-1738. b) Tsend, C. K. H.; Marquez, V. E.; Fuller, R. W.; Goldstein, B. M.; Arnett, G.; Hollingshead, M.; Driscoll, J. S. J. Med. Chem. 1989, 32, 1442-1446. c) Hasobe, M.; Mckee, J. G.; Borcherding, D. R.; Borchardt, R. T. Antimicrob. Agents Chemother. 1987, 31, 1849-1851.
- Paisley, S. D.; Wolfe, M. S.; Borchardt, R. T. J. Med. Chem. 1989, 32, 1415-1418.

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